



Poster 12

Capture and reduction of carbon emissions to maximize circularity in the steelmaking process



Azita Etminan

AUTHOR OF POSTER:
Azita Etminan

INSTITUTION:
Swansea University

OTHER AUTHORS:
Professor Peter Holliman, Swansea University
Professor Ian Mabbett, Swansea University
Dr Ciaran Martin, Tata Steel UK

ABSTRACT:

Emissions of carbon monoxide and carbon dioxide from Blast Furnace Gas, Basic Oxygen Furnace Gas, and Coke Oven Gas are significant sources. Shifting to a circular economy involves converting these gases into renewable fuels. This research introduces an innovative thermodynamic analysis, evaluating hydrogenating CO₂/CO-rich gases along with steel-off gases.

Process design: As synthetic natural gas can be used in integrated steelworks, substituting for natural gas, the direct carbon capture and utilisation using BFG and BOFG as a carbon source and H₂-rich COG as a hydrogen source is the focus of this research.

Thermodynamic analysis: The HSC Chemistry software is applied for process simulations of steel-off gas reforming without a catalyst. The study utilised a thermodynamic equilibrium analysis to identify optimal process parameters and reaction conditions. The objective was to enhance CO₂ hydrogenation and minimize carbon formation. The findings highlight variations in process gas characteristics, with the most effective operational parameters for maximising methane production involving Coke Oven Gas, CO₂-rich gas flow, and a specific temperature range.



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